**Water Review for final exam**

1. A water molecule is composed of \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ atoms.
2. Truncated c) liquid carbon dioxide e) none of the above
3. Forbidden d) three f) sewage
4. Water “sticks” to itself because\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ .
5. The O & H atoms have a partial charge **c)** the hydrogen sticks to a matrioska b.
6. They are glued together d) nuclear distribution package
7. The density of water is \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ .
8. 1 parsec/sec c) 1 g/ml
9. 1 kp/centigrade d) can never be determined since H exists in space
10. Ice is\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_dense than water.
11. More c) equal
12. Less d) question can never be answered
13. Water forms a “bubble” instead of laying flat when dropped on a surface because of \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ .
14. Electromagnetic spectrum c) hydrogen bonds
15. Radiation from the sun d) gamma radiation
16. An object with a density greater than 1 will\_\_\_\_\_\_\_\_\_\_\_\_\_\_ in water
17. Sink c) eat e) tip
18. Float d) list f) swim
19. There is/are \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ forms of ice.
20. 2 c) very few e) no
21. 1 d) many f) burning

Create a Diagram of 2 water molecules and showing a hydrogen bond